

APPENDIX

Gas exchange equation.

Gas exchange was simulated with a mass balance equation between alveolar gas space and tissue (Longobardo et al., 1996, Betzel et al., 2007). We assume to have one homogeneous alveolar compartment perfused by pulmonary blood flow. The gas exchange at inspiration and expiration will be:

Inspiration

$$\frac{d}{dt} \left((V_{Exp} + V_A(t)) \cdot F_{O_2 alv}(t) + Vap(t) \cdot C_{O_2 ap}(t) \right) = \dot{V}_A(t) \cdot F_{O_2 I} + (1 - ps) \cdot (Qpa(t) \cdot C_{O_2 ap}(t) - Qpv(t) \cdot C_{O_2 vp}(t))$$

Expiration

$$\frac{d}{dt} \left((V_{Insp} + V_A(t)) \cdot F_{O_2 alv}(t) + Vap(t) \cdot C_{O_2 ap}(t) \right) = -\dot{V}_A(t) \cdot F_{O_2 alv}(t) + (1 - ps) \cdot (Qpa(t) \cdot C_{O_2 ap}(t) - Qpv(t) \cdot C_{O_2 vp}(t)) \quad (1)$$

Where Vap is the pulmonary arterial volume, V_{Exp} (V_{Insp}) is the alveolar volume at the end of expiration (inspiration), V_A is the incremental alveolar volume, \dot{V}_A is the alveolar ventilation over time calculated from dV_{lungs}/dt in equation (4), subtracting the dead space ventilation calculated from equation (12). Qpa (Qpv) is the pulmonary arterial (venous) blood flow, ps is the pulmonary shunt, $C_{O_2 ap}$ ($C_{O_2 vp}$) is the O_2 concentration in the arterial (venous) pulmonary blood, $F_{O_2 alv}$ ($F_{O_2 I}$) is the molar fraction of O_2 in the alveoli (inspired air). Since ventilator flows and volumes are reported in BTPS units while blood gasses concentrations are expressed in STPD units a conversion needs to be made:

$$V_{BTPS} = \frac{P_{STPD} \cdot T_{BTPS}}{P_{BTPS} \cdot T_{STPD}} \cdot V_{STPD} = \frac{863}{(P_{amb} - P_{H_2O})} \cdot V_{STPD} \quad (2)$$

Using Dalton's law we can express the molar fraction in terms of partial pressure:

$$F_{O_2} = \frac{P_{O_2}}{P_{amb} - P_{H_2O}} \quad (3)$$

This leads to the following equations:

Inspiration

$$\frac{d}{dt} \left((V_{Exp} + V_A(t)) \cdot P_{O_2 alv}(t) + 863 \cdot Vap(t) \cdot C_{O_2 ap}(t) \right) = \dot{V}_A(t) \cdot P_{O_2 I} + 863 \cdot (1 - ps) \cdot (Qpa(t) \cdot C_{O_2 ap}(t) - Qpv(t) \cdot C_{O_2 vp}(t))$$

Expiration

$$\frac{d}{dt} \left((V_{Einsp} + V_A(t)) \cdot P_{O_2 alv}(t) + 863 \cdot Vap(t) \cdot C_{O_2 ap}(t) \right) = -\dot{V}_A(t) \cdot P_{O_2 alv}(t) + 863 \cdot (1 - ps) \cdot (Qpa(t) \cdot C_{O_2 ap}(t) - Qpv(t) \cdot C_{O_2 vp}(t)) \quad (4)$$

Solving the differential equation and considering that:

$$\begin{aligned} \frac{dV_A(t)}{dt} &= \begin{cases} \dot{V}_A(t) & \text{for inspiration} \\ -\dot{V}_A(t) & \text{for expiration} \end{cases} \\ \frac{dVap(t)}{dt} &= (1 - ps) \cdot Qvp(t) - Qap(t) \end{aligned} \quad (5)$$

We finally obtain:

Inspiration

$$\left((V_{Eexp} + V_A(t)) + 863 \cdot Vap(t) \cdot \frac{dC_{O_2 ap}(t)}{dP_{O_2 alv}(t)} \right) \cdot \frac{dP_{O_2 alv}(t)}{dt} = 863 \cdot (1 - ps) \cdot Qpv(t) \cdot (C_{O_2 ap}(t) - C_{O_2 vp}(t)) + \dot{V}_A(t) \cdot (P_{O_2 I}(t) - P_{O_2 alv}(t))$$

Expiration

$$\left((V_{Einsp} + V_A(t)) + 863 \cdot Vap(t) \cdot \frac{dC_{O_2 ap}(t)}{dP_{O_2 alv}(t)} \right) \cdot \frac{dP_{O_2 alv}(t)}{dt} = 863 \cdot (1 - ps) \cdot Qpv(t) \cdot (C_{O_2 ap}(t) - C_{O_2 vp}(t)) \quad (6)$$

We assume that O₂ and CO₂ partial pressures are equal in the alveoli and in the arterial blood. Therefore, $dC_{O_2 ap}/dP_{O_2 alv}$ is the slope of the dissociation curve in the operating point (Chiari et al., 1997). We considered a constant value of 0.0001625 (ml/mmHg·ml) (Longobardo et al., 1966). For the carbon dioxide similar equations can be used with the difference that $P_{O_2 I}$ can be neglected. The slope of the CO₂ dissociation curve is set to 0.0065 ml/(mmHg·ml) (Longobardo et al., 1966).